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A novel chlorinated norsesquiterpenoid and two related new metabolites from the soft coral *Paralemnalia thyrsoides*

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Abstract—A structurally unique chlorinated norsesquiterpenoid, paralemnolin A (1), along with two new metabolites, paralemnolin B (2) and C (3), have been isolated from the soft coral *Paralemnalia thyrsoides*. Their structures were determined by extensive spectroscopic analyses. Single-crystal X-ray diffraction analysis of 1 further establish the absolute stereochemistry of 1. © 2005 Elsevier Ltd. All rights reserved.

Soft coral of the genus *Paralemnalia* have been found to be a rich source of bioactive secondary metabolites. ^{1–4} During the course of our investigation on the bioactive chemical constituents from marine invertebrates, ^{5–7} two new norsesquiterpenoids (1 and 2), and one sesquiterpenoid (3) have been isolated from the soft coral *Paralemnalia thyrsoides*, collected by hand using scuba at Green Island, located off the southeast coast of Taiwan. We describe herein the structure elucidation of these compounds, and also the proposed biosynthesis pathway for the chlorinated metabolite 1 (Fig. 1).

The organism of *P. thyrsoides* was frozen immediately after collection and the freeze-dried organism was extracted sequentially with EtOAc. The EtOAc extract (33.0 g) was fractionated by open column chromatography on silica gel using *n*-hexane, *n*-hexane/EtOAc, and EtOAc/MeOH mixtures of increasing polarity. A fraction eluted with EtOAc/*n*-hexane (1:15) was further purified by normal-phase HPLC using EtOAc/*n*-hexane (1:50) to afford compound 1 (20.9 mg). Compounds 2

Keywords: Paralemnolin A; Paralemnolin B; Paralemnolin C; Paralemnalia thyrsoides; Soft coral.

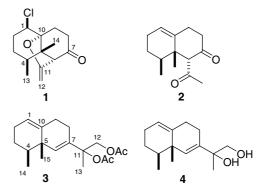


Figure 1. Structures of metabolites 1-4.

(27.3 mg) and 3 (35.5 mg) were eluted with EtOAc/ n-hexane (1:10) and further purified by normal-phase HPLC using EtOAc/n-hexane (1:30).

ESIMS of paralemnolin A (1)⁸ exhibited two ion peaks at m/z 277 [M+Na]⁺/279 [M+2+Na]⁺ (in a ratio ca. 1:0.3) in accordance with the presence of one chlorine atom (Table 1). A molecular formula of $C_{14}H_{19}ClO_2$ for 1 was established by HRESIMS, indicating five degrees of unsaturation. The NMR spectral data suggested the presence of one ketone (δ_C 206.6 qC), one 1-alkyl,

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Table 1. ¹H and ¹³C NMR spectral data of compounds 1–3

C/H	1			2			3		
	¹³ C ^a		¹ H ^b	¹³ C ^a		¹ H ^b	¹³ C ^a		¹ H ^b
1	60.0 d ^c		4.28 t (3.0) ^d	124.3 d ^c		5.67 m	120.3 d ^c		5.34 br s
2	30.6 t		1.92 m	25.7 t		1.99 m	25.4 t		1.98 m
			2.27 m			2.13 m			
3	23.3 t		1.39 m	27.2 t		1.39 m	27.4 t		1.48 m
			1.69 m			1.43 m			
4	33.5 d		1.71 m	35.1 d		1.71 m	37.0 d		1.50 m
4 5	49.2 s			46.7 s			38.4 s		
6 7	65.6 d		3.10 s	75.0 d		3.89 s	131.4 d		$5.62 d (2.7)^{d}$
7	206.6 s			207.3 s			135.2 s		
8	34.1 t	α	2.77 m	38.3 t	α	2.86 m	27.4 t		1.88 m
		β	2.35 m		β	2.29 m			2.16 m
9	32.9 t	α	2.01 m	31.7 t	α	2.56 m	30.0 t	α	2.10 m
		β	2.65 m		β	2.64 m		β	2.29 m
10	87.2 s			136.3 s			141.8 s		
11	158.8 s			204.4 s			82.9 s		
12	84.4 t		4.01 d (2.1)	33.9 q		2.22 s	68.1 t		4.15 d (11.1)
			4.32 d (2.1)						4.24 d (11.4)
13	16.4 q		0.85 d (6.6)	16.1 q		$0.93 \text{ d} (6.9)^{\text{d}}$	21.5 q		1.59 s
14	13.1 q		1.04 s	20.0 q		0.95 s	15.9 q		0.93 d (5.7)
15							20.8 q		0.93 s
OAc							170.9 s		
							21.2 q		2.08 s
							169.7 s		
							22.1 q		2.01 s

^a Spectra recorded at 75 MHz in CDCl₃ at 25 °C.

1-alkoxy-disubstituted carbon–carbon double bond ($\delta_{\rm C}$ 158.8 qC and 84.4 CH₂; $\delta_{\rm H}$ 4.01 and 4.32, each d, J=2.1 Hz), one oxygenated quaternary carbon ($\delta_{\rm C}$ 87.2), and one secondary alkyl chloride ($\delta_{\rm C}$ 60.0; $\delta_{\rm H}$ 4.28, dd, J=3.0, 3.0 Hz). The above functionalities also account for two of the five degrees of unsaturation, suggesting a tricyclic structure in 1.

The gross structure of 1 was established by 2D NMR spectroscopic analyses. From ¹H-¹H COSY spectrum of 1, it was possible to establish the proton sequences by the following cross-peaks: H-1/H₂-2, H₂-2/H₂-3; H₂-3/H-4, H-4/H₃-13, H₂-8/H₂-9 (Fig. 2). Its HMBC spectrum showed many informative correlations, such

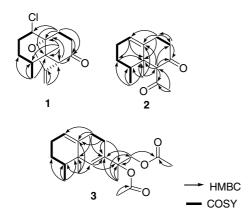


Figure 2. Selective ¹H-¹H COSY and HMBC correlations of 1-3.

as H-6/C-7 and C-10; H_2 -8/C-7 and C-10; H_2 -9/C-1, C-5, C-7, C-8, and C-10; H_2 -12/C-11 and C-6; H_3 -13/C-3, C-4, and C-5; and H_3 -14/C-4, C-5, C-6, and C-10 (Fig. 2). Thus, the unusual planar structure of **1** with an α -methylene THF structural unit was established unambiguously.

The relative stereochemistry of **1** was elucidated from the NOE interactions observed in a NOESY experiment (Fig. 3). In the NOESY experiment of **1**, H₃-14 was found to show NOE interactions with H-6, H₃-13, and one proton of H₂-9 ($\delta_{\rm H}$ 2.65, m), but not with H-1 and H-4. Furthermore, the other proton of H₂-9 ($\delta_{\rm H}$ 2.01, m) showed a NOE interaction with H-1, suggesting that H₃-14, H₃-13, and H-6 should be positioned on the same face and arbitrarily assigned as β -protons, and H-1 and H-4 were assigned as α -protons. To confirm the structure of **1** and establish its absolute configuration, a single-crystal X-ray crystallographic analysis was undertaken. The crystal structure (ORTEP diagram)

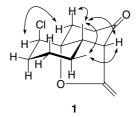


Figure 3. Key NOESY correlations of 1.

^b Spectra recorded at 300 MHz in CDCl₃ at 25 °C.

^c Multiplicity are deduced by HSQC and DEPT spectra and indicated by usual symbol.

^d J value (in Hz) in parentheses.

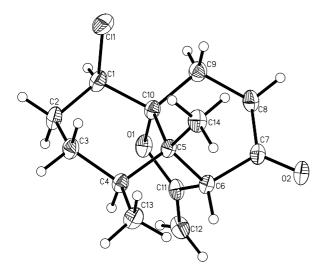


Figure 4. X-ray ORTEP diagram of 1.

is shown in Figure 4. The absolute stereochemistry was assigned on the basis of the Flack parameter, which refined to a value of 0.07(7).¹⁰ Thus, the structure of 1 was established unambiguously and the absolute configurations of C1, C2, C5, C6, and C10 were assigned as 1R, 4S, 5S, 6S, and 10R, respectively.

Compound 2^{11} was obtained as a white powder. Its HREIMS exhibited a molecular ion peak at m/z 220.1461 and established a molecular formula of $C_{14}H_{20}O_2$, corresponding to five degrees of unsaturation. The NMR data showed the presence of two ketones (δ_C 207.3 and 204.4, each qC), and one trisubstituted double bond (δ_C 124.3 CH, 136.3 qC). The above functionality account for three of the five degrees of unsaturation, suggesting a bicyclic structure in 2.

To establish the proton correlations in 2, the ¹H-¹H COSY spectrum was used to reveal the connectivities of H-1/H₂-2, H₂-2/H₂-3, H₂-3/H-4, H-4/H₃-13, and H₂-8/H₂-9 (Fig. 2). These data, together with the ¹H–¹³C long-range correlations observed in the HMBC experiment (Fig. 2), established the carbon skeleton of 2. The position of the acetyl group at C-6 was confirmed by the key HMBC correlations from H₃-12 to C-11 and C-6. The other ketone positioned at C-7 was confirmed by the following HMBC cross-peaks: H-6/C-5, C-7, C-8, and C-10; and H₂-8/C-7, C-9, and C-10. The above observations and other HMBC correlations as shown in Figure 2, established the planar structure of 2. Furthermore, the NOE correlations between H₃-14 and H_3 -13, and H_3 -14 and H-6 suggested the β -orientation of these protons. Thus, the structure of 2 was fully established.

Compound 3^{12} was isolated as a colorless oil. The HRE-SIMS of 3 established a molecular formula of $C_{19}H_{28}O_4$, implying six degrees of unsaturation. By comparison of the NMR spectral data of 3 with those of a known metabolite 4, 13 it was found that hydroxy groups attached to C-11 and C-12 in 4 (δ_{C-11} 75.1, δ_{C-12} 68.4,

Scheme 1. Proposed biosynthesis pathway of 1.

and $\delta_{\rm H}$ 3.42 and 3.61) were converted to acetoxy groups in 3 ($\delta_{\rm C-11}$ 82.9, $\delta_{\rm C-12}$ 68.1, $\delta_{\rm H}$ 4.15 and 4.24). The NOE correlations between H₃-15 and H₃-14 suggested these protons were located on the same face in the molecule of 3.

On the basis of a biosynthetic reaction suggested by Bowden et al., a biogenetic precursor 1(10)-aristolene (5) could be transformed into a diene intermediate (6). It was considered to be possible that oxidative cleavage at double bond of the isopropenyl group of 6 could afford metabolite 2 (Scheme 1). Furthermore, the proposed biosynthetic pathway of 1 and the biogenetic relationship between 1 and 2 was shown also in Scheme 1. This pathway involves initial chlorination at C-1 of 2 to form a carbonium ion with a positive charge at C-10. Subsequent enolization of the acetyl group and the following cyclization of the enol hydroxyl with C-10 would lead to the formation of 1.

Preliminary biological activity screening revealed that these three compounds are not active against the growth of a limited panel of cancer cell lines, including A549 (human lung carcinoma), HepG2 (human hepatocellular carcinoma), and MAD-MB-231 (human breast carcinoma) cells. The results of further biological activity screening will be reported elsewhere in the future.

Acknowledgments

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- 8. Paralemnolin A (1): colorless crystal; mp 96-98 °C; $[\alpha]_{\rm D}^{25}$ -133 (c 0.82, CHCl₃); IR (KBr) $\nu_{\rm max}$ 1722, 1674 cm⁻¹; ¹H NMR and ¹³C NMR data, see Table 1;

- ESIMS m/z (relative intensity) 277/279 $[M+Na]^+$ (1/0.3); HRESIMS m/z 277.0973 [M+Na]⁺ (calcd for C₁₄H₁₉ClO₂Na, 277.0971).
- 9. Crystallography data (excluding structure factors) of 1 have been deposited with the Cambridge Crystallographic Data Center as supplementary publication numbers CCDC280602. Copies of the data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK [fax: +44(0)-1223-336033 or e-mail: deposit@ccdc.cam.ac.uk].
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 Paralemnolin B (2): white powder; mp 47–49 °C; [α]_D²⁵ –600 (c 1.46, CHCl₃); IR (KBr) ν_{max} 1697 cm⁻¹; ¹H NMR and ¹³C NMR data, see Table 1; EIMS m/z 220 [M]⁺; HREIMS
- m/z 220.1461 [M]⁺ (calcd for $C_{14}H_{20}O_{2}$, 220.1463). 12. Paralemnolin C (3): colorless oil; $[\alpha]_D^{25}$ -142 (c 1.08, CHCl₃); IR (KBr) ν_{max} 1745 cm⁻¹; ¹H NMR and ¹³C NMR data, see Table 1; ESIMS m/z 343 [M+Na]⁺; HRESIMS m/z 343.1887 $[M+Na]^+$ (calcd for C₁₉H₂₈O₄Na, 343.1885).
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